# Development of New Schiff Base Derivatives as Fluorescent Chemosensors, NIR Emitters and Catalysts

A Thesis

Submitted for the Degree of

### DOCTOR OF PHILOSOPHY

In the School of Basic Sciences

By

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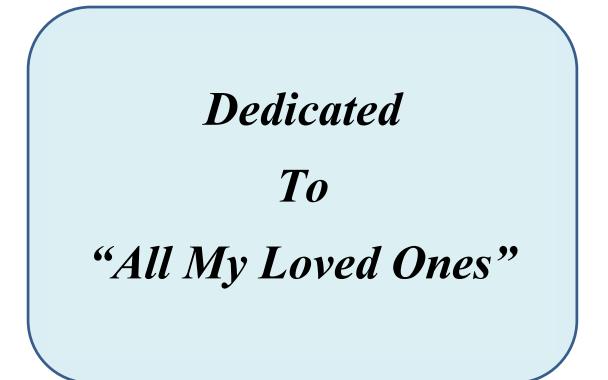
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## **Declaration by the Research Scholar**

I hereby declare that the entire work embodied in this Thesis is the result of investigations carried out by me in the *School of Basic Sciences,* Indian Institute of Technology Mandi, under the supervision of *Dr. Pradeep C. Parameswaran,* and that it has not been submitted elsewhere for any degree or diploma. In keeping with the general practice, due acknowledgements have been made wherever the work described is based on finding of other investigators.

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# **Declaration by the Research Advisor**

I hereby certify that the entire work in this Thesis has been carried out by *Mr. Abhishek Kumar Gupta*, under my supervision in the *School of Basic Sciences*, Indian Institute of Technology Mandi, and that no part of it has been submitted elsewhere for any Degree or Diploma.

Signature:

Name of the Guide: Dr. Pradeep C. Parameswaran Date:

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## ABBREVIATIONS

# Abbreviations

# Symbols

	•
Φ	Quantum Yield
K <sub>b</sub>	Binding Constant
$\lambda_{em}$	Position of the Emission Maximum
$\lambda_{ex}$	Position of the Excitation Maximum
3	Molar Extinction Coefficient
$ au_{avg}$	Average Life Time
δ	Chemical Shift
t <sub>1/2</sub>	Half Life Time
Eg	Energy Gap
	Solvents
CHCl <sub>3</sub>	Chloroform
CH <sub>2</sub> Cl <sub>2</sub> /DCM	Dichloromethane
DMF	Dimethylformamide
DMSO	Dimethylsulfoxide
EtOH	Ethanol
MeOH	Methanol
TEA	Triethylamine
THF	Tetrahydrofuran
	Chemicals
ATP	Adenosine-5'-triphosphate
СТР	Cytidine-5'-triphosphate
ADP	Adenosine diphosphate
AMP	Adenosine monophosphate
PPi	Potassium pyrophosphate
DFP	2,6-Diformylphenol
TRIS	Tris(hydroxymethyl)aminomethane
РА	Phenyl acetate

ABBREVIATIONS		
BnOH	Benzyl alcohol	
BnA	Benzyl acetate	
EA	Ethyl acetate	
IA	Isopropyl acetate	
НА	Hexyl acetate	
BuA	Butyl acetate	
СО	Canola oil	
МО	Methyl oleate	
ML	Methyl linoleate	
FAME	Fatty acid methyl ester	
AIBN	Azobis(isobutyronitrile)	
HEPES	4-(2-Hydroxyethyl)-1-piperazineethanesulfonic acid	
BODIPY	Boron dipyrromethane difluoride	
PMMA	Poly(methylmethaacrylate)	
4-MTA	4-Methylthioaniline	
Bu <sub>4</sub> NPF <sub>6</sub>	Tetrabutylammonium hexafluorophosphate	
	Mechanisms	
AIE	Aggregation Induced Emission	
СТ	Charge Transfer	
ESIPT	Excited State Intramolecular Proton Transfer	
FRET	Fluorescence Resonance Energy Transfer	
ICT	Intramolecular Charge Transfer	
MLCT	Metal Ligand Charge Transfer	
eT	Electron Transfer	
ET	Energy Transfer	
PET	Photo-induced Electron Transfer	
ILCT	Intra-Ligand Charge Transfer	
Instruments		
<sup>13</sup> C NMR	Carbon Nuclear Magnetic Resonance	
<sup>1</sup> H NMR	Proton Nuclear Magnetic Resonance	

ABBREVIATIONS	
<sup>31</sup> P NMR	Phosphorus Nuclear Magnetic Resonance
HR-MS	High Resolution Mass Spectrometry
ESI-MS	Electron Spray Ionization Mass Spectrometry
FT-IR	Fourier Transform Infrared Spectroscopy
TEM	Transmission Electron Microscopy
TGA	Thermogravimetric Analysis
SCXRD	Single Crystal X-Ray Diffraction
CV	Cyclic Voltammetry
GC	Gas Chromatography
DLS	Dynamic Light Scattering
DRS	Diffuse Reflectance Spectroscopy
UV-Vis	Ultraviolet-Visible
AAS	Atomic Absorption Spectroscopy
FID	Flame Ionized Detector
	Others
Ι	Fluorescence intensity in the presence of analyte
Io	Fluorescence intensity in the absence of analyte
НОМО	Highest Occupied Molecular Orbital
HPLC	High Performance Liquid Chromatography
LUMO	Lowest Unoccupied Molecular Orbital
NIR	Near Infra-Red
TLC	Thin Layer Chromatography
TMS	Tetramethylsilane
μΜ	Micro-molar
mM	Milli-molar
nM	Nano-molar
Ns	Nano-second
Κ	Keto
Е	Enol
°C	Degree Celsius

ABBREVIATIONS	
m.p.	Melting Point
CCDC	Cambridge Crystallographic Data Center
DFT	Density Functional Theory
CIFs	Complexation Induced Shifts
DL	Detection Limit
PPM	Parts Per Million
HRLs	Hydroxyl Rich Ligands
pН	Potential of Hydrogen
EDG	Electron Donating Group
EWG	Electron Withdrawing Group
ORTEP	Oak Ridge Thermal-Ellipsoid Plot

## Abstract

Schiff base molecules are gaining attention in recent years because of their ability to form compounds that are relevant to diverse areas. Schiff bases possess excellent characteristics including structural similarities with biological substances, relatively simple synthetic procedures and synthetic flexibility that allow structural design as per the requirements. Schiff bases often act as excellent chelating agents and form a variety of complexes with transition metal ions. Because of these unique properties, Schiff bases are widely used in the development of molecular sensors, catalysts, molecular switches, optical data storage devices and bio-mimetic compounds. However, there are still many areas where Schiff bases have not yet been fully explored. For example, the development of chemodosimeters for the selective detection of cations, especially in presence of interfering cations, is still a challenging task. The selective detection of biological macro anions like adenosine triphosphate (ATP), cytidine triphosphate (CTP) etc. is also a difficult task because of the larger size of such anions. Similarly, studies on Schiff base derivatives as solid state near infra-red (NIR) emitters and multifunctional materials are scarce in the literature.

In the present thesis, we have tried to address some of the above issues by developing suitable Schiff base molecular systems. We have developed a pyrene based Schiff base chemodosimeter for the selective detection of Nb<sup>5+</sup> ions in mixed aqueous media in presence of interfering cations and a series of hydroxyl-rich Schiff base receptors for the selective detection of macro anions like ATP and CTP in 100% aqueous environments. In continuation, we have developed a series of multifunctional zinc complexes for photo luminescent and catalytic applications using hydroxyl-rich compartmental Schiff base ligands. The photoluminescence properties of these zinc complexes were explored in solid state, solutions and in polymer matrix, which revealed their good potential as tunable solid state emitters. Some of these complexes acted as efficient catalysts for the transesterification of simple esters as well as vegetable oils revealing their potential in biodiesel generation. Finally, we have developed an extended piconjugated compartmental Schiff base molecule built on 2, 6-diformyl phenol derivatives that acted as deep red to NIR solid state emitter based on excited state intramolecular proton transfer (ESIPT) mechanism.